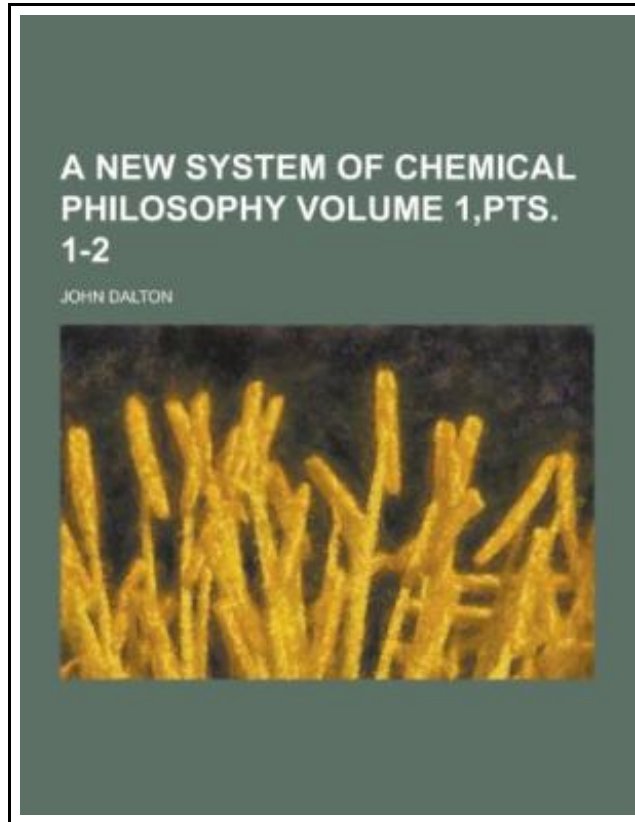


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Theclassics.us, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1808 edition. Excerpt: . They seem to think that the acid is decomposed in this case: but they have not advanced any opinion, that either fluoric or muriatic acid gas consists entirely of hydrogen and oxygen. Section 2. OXYGEN WITH AZOTE. The compounds of oxygen with azote, hitherto discovered, are five; they may be distinguished by the following names; nitrous gas, nitric acid, nitrous oxide, nitrous acid, and oxynitric acid. In treating of these, it has been usual to begin with that which contains the least oxygen, (nitrous oxide) and to take the others in order as they contain more oxygen. Our plan requires a different principle of arrangement; namely, to begin with that which is most simple, or which consists of the smallest number of elementary particles, which is commonly a binary compound, and then to proceed to the ternary and other higher compounds. According to this principle, it becomes necessary to ascertain, if possible, whether any of the above, and which of them, is a binary compound. As far as the specific gravities of the two simple gases are indicative of the weights of their atoms, we should conclude that an atom of azote is to one of oxygen as 6 to 7 nearly; the relative weights of ammonia and water also give countenance (o such a ratio. But the best criterion is derived from a comparison of the specific gravities of the compound gases themselves. Nitrous gas has the least specific gravity of any...



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